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Structural evolution across the metal-insulator transition of strongly distorted $Lu_{1-x}Sc_xNiO_3$ perovskites (x = 0, 0.1, 0.2)

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Abstract

RNiO₃ perovskites have been described to present thermally driven metal-insulator transitions (at T_{MI}) as a function of the rare-earth ion size (R = Pr to Lu). Aiming to extend the stability range of RNiO₃ for smaller R³⁺ ions, we prepared Lu_{1-x}Sc_xNiO₃ (x = 0, 0.1, 0.2) perovskites, being Sc³⁺ ions substantially smaller than Lu³⁺, by using a multi-anvil high-pressure synthesis device at 10 GPa. We have studied the structural evolution of Lu_{0.9}Sc_{0.1}NiO₃ by synchrotron x-ray diffraction (SXRD) from room temperature to 350 °C. The symmetry of the lattice evolves from monoclinic (*P*₂₁/*n*) to orthorhombic (*Pbnm*) upon heating across T_{MI} (\approx 320 °C), with the existence of two chemically and crystallographically distinct nickel sites in the insulating, monoclinic regime, whereas the metallic phase has a single NiO₆ environment. A simultaneous structural and electronic transition implies an abrupt evolution of the lattice parameters and size of the NiO₆ octahedra upon entering the metallic regime, leading to the merging of the disproportionated Ni-O bond lengths. The magnetic properties correspond to the establishment of antiferromagnetic correlations at the Ni sublattice; a decrease of the T_N ordering temperature from 122 K (x = 0) to 113 K (x = 0.2) is observed as the Sc content increases, which is concomitant with a more distorted perovskite structure.

Introduction

Rare-earth nickelates RNiO₃ have attracted the interest of researchers over the last three decades aiming to establish the relationships between structure and observed physical properties. While initially described in 1971 by Demazeau *et al* [1] the materials underwent a resurgence of interest after the discovery of a thermally driven metal-insulator (MI) transition depending on the rare-earth ion size [2–6]. In the case of small rare-earths (R = Ho-Lu, Y) it was subsequently shown that the symmetry of the lattice evolves from monoclinic (P2₁/n) to orthorhombic (*Pbnm*) upon heating across T_{MI} [7–11], with the presence of two nickel sites from the chemical and crystallographical point of view in the insulating, monoclinic regime, whereas the metallic phase has a single NiO₆ environment. The phenomenology involving MI transitions is peculiar of electron-correlated systems, and therefore RNiO₃ oxides have always been taken as paradigmatic examples where to investigate fundamental problems of condensed matter [12]. Moreover, their properties find applications as multiferroic oxides [13] superlattices [14–16], fuel cells [17] memory devices [18] or bio-electronic interfaces [19]. Negative magnetorresistance has been described in thin films of NdNiO₃ [20, 21], and superconductivity below $T_C = 9-15$ K for infinite-layer derivatives of composition Nd_{0.8}Sr_{0.2}NiO₂ [22]. First-principles electronic structure calculations show that charge ordered rare-earth nickelates of the type RNiO₃ (R = Ho, Lu, Pr and





Nd) are multiferroic with very large magnetically-induced ferroelectric (FE) polarizations [13]; multiferroicity has also been described in transition-metal perovskites derived from RNiO₃, like R₂NiMnO₆ [23, 24].

Since the first diffraction studies across the metallic-insulator transition [3–5], the observed symmetry reduction in the insulating regime [7–9] was ascribed to a charge disproportionation effect, or uncomplete charge-ordering phenomenon, where Ni³⁺ (metallic regime) evolves into Ni^{3+ δ} and Ni^{3- δ} (insulating regime) located in two differently-sized NiO₆ octahedra. It is owing to the asymmetric distribution of two distinct Ni1 and Ni2 atoms coordinated in octahedra with distinct Ni–O distances [25–27]. Alternatively, the structural distortion has been interpreted as the result of a bond disproportionation: the d⁸ electronic configuration is preserved in Ni atoms, whereas a positive charge is segregated at the oxygen positions, according to Ni-3*d* and O-2*p* hybridization. A configuration 3d⁸L¹ \rightarrow 3d⁸ + 3d⁸L² is ascribed to the disproportionated state, according to spectroscopic studies [26, 28, 29].

The stabilization of these materials containing Ni^{3+} becomes more difficult with the reduction in size of the rare-earth cation along the lanthanide series. After the successful preparation of RNiO₃ for the smallest rare-earths R = Er, Tm, Yb, Lu, at moderate pressures (2 GPa), we were able to study by neutron powder diffraction (NPD) the crystal evolution across T_{MI} (for instance, [25] for Yb and Tm, or [10] for LuNiO₃). Aiming to extend the stability range of RNiO₃ for even smaller R³⁺ ions, we proposed the preparation of Lu_{1-x}Sc_xNiO₃ perovskites, being Sc³⁺ ion (VIII: 0.870 Å) substantially smaller than Lu³⁺ (VIII: 0.977 Å). This involves a decrease of the tolerance factor, with an additional tilting effect of the octahedra and implying even more severe preparation conditions. This was achieved in a multianvil high-pressure synthesis device, able to reach over 10 GPa.

Moreover, these compounds exhibit antiferromagnetic magnetic structures at low temperature, arising from competing interactions at the Ni sublattice, with Néel temperatures identical to T_{MI} for larger R cations (R = Pr, Nd) whereas a lower T_N is described for R = Sm \rightarrow Lu [30–33]. Additionally, some magnetic rareearths become long-range ordered, sometimes with different periodicity, as observed for HoNiO₃ [32].

This work describes the preparation of these novel, extremely distorted, nickel perovskites, and the structural investigation across the MI transition of the x = 0.1 specimen, showing the evolution from the room-temperature (RT) monoclinic and insulating phase to the high-temperature orthorhombic, metallic phase. High angular resolution SXRD is required since the metric of the monoclinic unit-cell at RT involves monoclinic β angles below 90.15°, implying a strong pseudo-orthorhombic character. The magnetic properties display a reduction of the Néel temperature, which is also consistent with the increment of the structural distortion observed by diffraction methods.



Figure 2. Rietveld plots from SXRD data of Lu_{0.9}Sc_{0.1}NiO₃ at (a) 25 °C and (b) 350 °C, after refinement in P2₁/n and Pbnm space groups, respectively. The insets show the characteristic monoclinic splitting between (-224) and (224) reflections, which vanishes above T_{MI}. Discrepancy factors (a) at 25 °C: $R_p = 10.5\%$, $\chi^2 = 4.84$, $R_{Bragg} = 9.73\%$ (b) at 350 °C: $R_p = 10.7\%$, $\chi^2 = 2.75$, $R_{Bragg} = 10.60\%$.

Experimental

Polycrystalline samples of $Lu_{1-x}Sc_xNiO_3$ (x = 0, 0.1, 0.2) were prepared under high-pressure conditions in a Kawai-type multianvil module (Max Voggenreiter GmbH). The x = 0 perovskite, LuNiO₃, was prepared for comparative purposes. The starting materials (0.5-x/2)Lu₂O₃, (x/2)Sc₂O₃, and Ni(OH)₂ in the stoichiometry ratio are thoroughly mixed with ~30 wt.% KClO₄, which serves as the oxidizing agent. The precursor powders were placed in a gold capsule, sealed and set in a cylindrical graphite heater. All these sample assembly was contained in a semi-sintered octahedron made of Ceramacast 584-OF. The pressure was generated by compressing the octahedron (edge length 14mm) with eight pieces of WC anvils with a truncated edge length of 8 mm. For each synthesis, the sample was subjected to heat treatment at 900 °C for 20 min under 10 GPa. The temperature was quenched to room temperature before releasing pressure slowly. The resultant products were washed with water to dissolve KCl and then dried in air at 100 °C for 2 h. Phase purity of the obtained $Lu_{1-x}Sc_xNiO_3$ (x = 0, 0.1, 0.2) polycrystalline samples was assessed by x-ray powder diffraction (XRD) at room temperature with CuK α radiation.

SXRD patterns were collected in the powder diffraction station of the MSPD beamline at the ALBA synchrotron, Barcelona (Spain), with 38 keV energy, $\lambda = 0.3252$ Å, and with the high angular resolution MAD set-up [34]. A selected sample Lu_{0.9}Sc_{0.1}NiO₃ was contained in a quartz capillary of 0.5 mm diameter. The acquisition temperatures were 25 °C (RT) and 100, 150, 200, 250, 280, 300, 320 and 350 °C. The refinement of the structures was performed by the Rietveld method. DC magnetic susceptibility was measured with a commercial Magnetic Property Measurements System (MPMS-III, Quantum Design) in the temperature range 2 K ~ 300 K under an external magnetic field of 0.5 T after field-cooled from RT.

Results and discussion

Figure 1(a) shows the laboratory XRD patterns of $Lu_{1-x}Sc_xNiO_3$ (x = 0.0, 0.1, 0.2) polycrystalline raw samples. The peak marked by an asterisk is the main peak of KCl coming from KClO₄, which would be washed away before characterization. Lattice parameters obtained from the Rietveld refinements from laboratory XRD patterns are illustrated in figure 1(b). With increasing x in the $Lu_{1-x}Sc_xNiO_3$ series, *a* and *c* decrease, but *b* increases with a net decrease of unit-cell volume V, in line with the fact that Sc^{3+} has a smaller ionic radius than Lu^{3+} .

The crystal structure of Lu_{0.9}Sc_{0.1}NiO₃ at RT, in the insulating region, was refined in the monoclinic $P2_1/n$ space group. Above T_{MI}, the structure was defined in the orthorhombic *Pbnm* space group; the abrupt variation of the unit-cell parameters, particularly the monoclinic beta angle, described below, indicated T_{MI} = 320 °C, slightly lower than that of LuNiO₃, of 326 °C [10]. The SXRD patterns after the Rietveld refinement at RT and 350 °C are displayed in figure 2. The plots at the remaining temperatures are gathered in figure S1 is available online at stacks.iop.org/MRX/7/126301/mmedia of the Supplementary Information. Thanks to the excellent crystallinity and the high angular resolution of the MSPD diffractometer, it was possible to resolve certain characteristic peak splittings in the monoclinic phase, as illustrated in the insets of figure 2, for the (-2 2 4) and (2 2 4) reflections, that merge into a single peak above T_{MI}.



Figure 3. Temperature variation of the unit-cell parameters and volume across the MI transition ($T_{MI} = 320$ °C). Immediately above T_{MI} the symmetry becomes orthorhombic, and the monoclinic angle abruptly falls to 90°.

Table 1. (a) Structural parameters for Lu_{0.9}Sc_{0.1}NiO₃ from SXRD data at 25 °C, sg $P2_1/n$; a = 5.1086(1), b = 5.4971(1), c = 7.3397(2) Å, β = 90.146 (2)°. (b) Structural parameters for Lu_{0.9}Sc_{0.1}NiO₃ from SXRD data at 350 °C, s.g. *Pbnm*, a = 5.1279(2), b = 5.5249(2), c = 7.3386(2) Å.

Atom	site	х	у	Z	$B(\text{\AA}^2)$
Lu,Sc	4e	0.9786(3)	0.0777(2)	0.2514(5)	0.75(3)
Ni1	2d	0.5	0.0	0.0	0.89(7)
Nil	2 <i>c</i>	0.5	0.0	0.5	0.89(7)
O1	4e	0.0971(2)	0.4798(2)	0.2584(4)	0.1(2)
O2	4e	0.6912(4)	0.2919(4)	0.0609(2)	0.7(3)
O3	4 <i>e</i>	0.1787(4)	0.1856(5)	0.9805(3)	2.1(6)
(b)					
Lu,Sc	4 <i>c</i>	0.9785(4)	0.0778(2)	0.25	1.21(6)
Ni1	4b	0.5	0.0	0.0	1.11(9)
01	4 <i>c</i>	0.0918(3)	0.4886(3)	0.25	0.9(4)
O2	8 <i>d</i>	0.6883(3)	0.3014(3)	0.0341(2)	2.2(4)

Therefore, the crystal structure at 25 °C and below the transition temperature is defined in the $P2_1/n$ symmetry, as proposed for RNiO₃ perovskites [7], with characteristic unit-cell parameters $a \approx \sqrt{2} a_0$, $b \approx \sqrt{2} a_0$ and $c \approx 2a_0$, where a_0 defines the unit-cell of the simple cubic perovskite. In this structure Ni1 and Ni2 are located at 2*d* and 2*c* sites, respectively, and O1, O2 and O3 oxygen atoms at 4*e* Wyckoff sites. Table 1(a) lists the main crystallographic parameters at RT. In the crystal structure there are small Ni1O₆ and large Ni2O₆ octahedra alternating along the three direction, as displayed in figure 1(*c*). This arrangement has been interpreted as a charge disproportionation effect [7–11, 25]. It is remarkable the small monoclinic β angle at RT, of 90.15°, indicating a strong pseudo-orthorhombic character. This value compares with those reported in *R*NiO₃ (R = Ho, Y, Er and Lu) perovskites, with β angles ranging from 90.08° for R = Y, Ho to 90.16° for LuNiO₃ at RT [9].

Above T_{MI} , in the metallic region, the crystal framework corresponds to the standard *Pbnm* orthorhombic superstructure of perovskite, with a single Ni atom at 4*b* sites and two O1 and O2 atoms at 4*c* and 8*b* Wyckoff positions, respectively (table 1(b)). This is the conventional description of the GdFeO₃-type perovskite structure [1].

The variation of the structural parameters across the phase transition at $T_{MI} = 320$ °C has been investigated from SXRD data (figure 3). The *a* and *b* unit-cell parameters regularly increase in the measured temperature interval; a conspicuous contraction of *c* parameter is realized upon entering the metallic regime across the phase transition. Interestingly, while a decrease of *b* was described for large R perovskites (R = Pr, Nd, Sm) below T_{MI} [27], an increment of the *b* lattice parameter occurs for RNiO₃ with smaller rare earths (R = Ho, Y, Er, Lu) [10].

It is interesting to compare the evolution of the interatomic Ni–O bonds between the monoclinic and the orthorhombic phases. Table 2 included the main Ni–O interatomic distances for the two types of NiO₆ octahedra in the $P2_1/n$ structure (at 25 °C), and the single type of NiO₆ octahedron at the *Pbnm* s.g. (at 350 °C). The average $\langle Ni-O \rangle$ distances indicate that, in the monoclinic model, Ni1O₆ octahedron is significantly smaller



 $\begin{array}{l} \textbf{Table 2. Main bond distances (Å) for monoclinic (at 25 °C)} \\ and orthorhombic (at 350 °C) Lu_{0.9}Sc_{0.1}NiO_3. For the \\ calculation of the bond valence sums (BVS) the following \\ parameters were used: B = 0.37, R_0(Lu^{3+}) = 1.971, \\ R_0(Sc^{3+}) = 1.849, R_0(Ni^{2+}) = 1.654 \, [35]. \end{array}$

T(°C)	25		350
Ni1-O1(x2)	1.846(3)	Ni-O1(x2)	1.895(4)
Ni1-O2(x2)	1.930(2)	Ni-O2(x2)	1.9411(16)
Ni1-O3(x2)	1.938(3)	Ni-O2'(x2)	1.9548(16)
(Ni1–O)	1.905(1)	$\langle Ni-O \rangle$	1.9304(5)
BVS	3.41(1)		3.16(1)
Ni2-O1(x2)	1.962(3)		
Ni2-O1(x2)	2.000(2)		
Ni2-O1(x2)	1.960(3)		
(Ni2–O)	1.974(1)		
BVS	2.81(1)		
(Ni–O)	1.939(2)		

 $(\langle Ni1-O \rangle = 1.905(1) \text{ Å})$ than Ni2O₆ octahedron $(\langle Ni2-O \rangle = 1.974(1) \text{ Å})$. This is highlighted in figure 4, illustrating the alternation of both types of contracted and expanded octahedra. The three different Ni–O bond lengths existing at each type of Ni octahedron, in $P2_1/n$, converge to three unique Ni–O distances in the orthorhombic-metallic phase, suggesting that there is a disappearance of the charge disproportionation at the MI transition.

The valences of the cations and anions, BVS = ΣS_{ij} , were estimated using Sij = exp[(R₀-R_{ij})/B], corresponding to the Brown's Bond-Valence Model [35]. In the monoclinic phase, the BVS for Ni1 and Ni2 are 3.41+ and 2.81+, respectively, significantly above and below the nominal value of 3+, (table 2), indicating the mentioned charge disproportionation, Ni^{3+ δ} and Ni^{3- δ}. In average, the δ value is 0.30, as observed in other members for small rare-earth cations [9].

Figure 5(a) illustrates the thermal variation of the *dc* magnetic susceptibility $\chi(T)$ for these three samples. The antiferromagnetic (AFM) transition at T_N is clearly manifested as a kink in $\chi(T)$. This singularity broadens up with increasing of the Sc content, probably due to the enhanced lattice distortion with Sc doping. It also



shows that T_N decreases linearly with x in $Lu_{1-x}Sc_xNiO_3$ series; a decrease of the T_N ordering temperature from 122 K (x = 0) to 113 K (x = 0.2) is observed, which is connected with a more distorted perovskite structure with reduced superexchange Ni–O-Ni angles.

Former investigations in the RNiO₃ perovskite series described low-spin Ni(III) as ground state [32], also supported in further studies [33]. Moreover, Curie-Weiss fits unveil paramagnetic moments (sufficiently above the onset for magnetic ordering) suggesting a low-spin $3t_{2g}^6 e_g^1$ electronic configuration, S = 1/2, as illustrated for YNiO₃ [36]. In complement, the magnetic structures investigated by neutron diffraction at low temperatures disclose ordered magnetic moments compatible with $S = \frac{1}{2}$ [8, 37].

Figure 5(b) shows a complete RNiO₃ phase diagram with the novel T_N and T_{MI} points, corresponding to the smallest-sized R^{3+} members of the series. Based on the classical diagram published by Torrance *et al* [5], showing a divergence between the metal-insulator transitions, T_{MI} , and the antiferromagnetic ordering temperature, T_N , for rare-earth sizes slightly smaller than Nd³⁺, we add a somewhat smaller T_{MI} for Lu_{0.9}Sc_{0.1}NiO₃ than that observed for LuNiO₃, as well as reduced T_N 's for the Sc_{0.1} and Sc_{0.2} members, thus exhibiting the lowest T_N of the full RNiO₃ series, given the strongest distortion exhibited by these perovskites.

Conclusions

We have demonstrated that an increased perovskite distortion beyond LuNiO₃ is possible in the series $Lu_{1-x}Sc_xNiO_3$. Very high pressure of 10 GPa is required for the stabilization of the novel members. The samples present MI transitions with concomitant structural changes, evolving from a monoclinic phase below T_{MI} to an orthorhombic symmetry in the metallic high-T structure. A dramatic rearrangement of the unit-cell parameters is observed when approaching $T_{MI} = 320$ °C for x = 0.1. The monoclinic, low-T insulating phase contains two types of Ni octahedra, corresponding to a charge-disproportionation effect. The average charge disproportionation between Ni1 and Ni2 is about ± 0.3 , as observed in other RNiO₃ members with small rare-earth cations. The magnetic properties show an antiferromagnetic ordering with T_N decreasing as the Sc content increases, as corresponds to a more distorted perovskite structure.

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References

- [1] Demazeau G, Marbeuf A, Pouchard M and Hagenmuller P 1971 J. Solid State Chem. 3 582
- [2] Ahn CH, Triscone JM and Mannhart J 2003 Nature 424 1015
- [3] Torrance J B, Lacorre P, Asavaroengchai C and Metzger R M 1991 J. Solid State Chem. 90 168
- [4] Lacorre P, Torrance J B, Pannetier J, Nazzal A I, Wang P W and Huang T C 1991 J. Solid State Chem. 91 225
- [5] Torrance J B, Lacorre P, Nazzal A I, Ansaldo E and Niedermayer C 1992 Phys. Rev. B 45 8209
- [6] Catalan G 2008 Phase Transitions 81 729
- [7] Alonso J A, Martínez-Lope M J, Casais M T, Aranda M A G and Fernández-Díaz M T 1999 J. Am. Chem. Soc. 121 4754
- [8] Alonso J A, García-Muñoz J L, Fernández-Díaz M T, Aranda M A G, Martínez-Lope M J and Casais M T 1999 Phys. Rev. Lett. 82 3871
- [9] Alonso J A, Martínez-Lope M J, Casais M T, García-Muñoz J L and Fernández-Díaz M T 2000 Phys. Rev. B 61 1756
- [10] Alonso J A, Martínez-Lope M J, Casais M T, García-Muñoz J L, Fernández-Díaz M T and Aranda M A G 2001 *Phys. Rev.* B 64 094102
- [11] Fernández-Díaz M T, Alonso J A, Martínez-Lope M J, Casais M T, García-Muñoz J L and Aranda M A G 2000 Phys. B: Condens. Matter. 276–278 218
- [12] Medarde M L and 1997 J Phys. Condens. Matter. 9 1679
- [13] Giovannetti G, Kumar S, Khomskii D, Picozzi S and van den Brink J 2009 Phys. Rev. Lett. 103 156401
- [14] Wu M, Benckiser E, Audehm P, Goering E, Wochner P, Christiani G, Logvenov G, Habermeier H U and Keimer B 2015 Phys. Rev. B 91 195130
- [15] Bruno F Y et al 2013 Phys. Rev. B 88 195108
- [16] Catalano S, Kreisel J, Gibert M, Triscone J M, Íñiguez J and Fowlie J 2017 Reports Prog. Phys. 81 046501
- [17] Amow G and Skinner S J 2006 J. Solid State Electrochem. 10 538
- [18] Ramadoss K et al 2018 IEEE Electron Device Lett. 39 1500
- [19] Zhang H T et al 2019 Nat. Commun. 10 1651
- [20] Wang L, Ju S, You L, Qi Y, Guo Y, Ren P, Zhou Y and Wang J 2016 *Sci Rep.* **5** 18707
- [21] Stupakov A, Pacherova O, Kocourek T, Jelinek M, Dejneka A and Tyunina M 2019 *Phys. Rev.* 99 085111
- [22] Li D, Lee K, Wang B Y, Osada M, Crossley S, Lee H R, Cui Y, Hikita Y and Hwang H Y 2019 Nature 572 624–7
- [23] Katari V, Babu P D, Mishra S K, Mittal R, Bevara S, Achary S N, Deshpande S K and Tyagi A K 2016 J. Amer. Ceram. Soc. 99 499–506
- [24] Sayed F N, Achary S N, Jayakumar O D, Deshpande S K, Krishna P S R, Chatterjee S, Ayyub P and Tyagi A K 2011 J. Mater. Res. 26 567–77
- [25] Alonso J A, Martínez-Lope M J, Presniakov I A, Sobolev A V, Rusakov V S, Gapochka A M, Demazeau G and Fernández-Díaz M T 2013 Phys. Rev. B 87 184111
- [26] Mercy A, Bieder J, Íñiguez J and Ghosez P 2017 Nat. Commun. 8 1677
- [27] Gawryluk D J, Klein Y M, Shang T, Sheptyakov D, Keller L, Casati N, Lacorre P, Fernández-Díaz M T, Rodríguez-Carvajal J and Medarde M 2019 Phys. Rev. B 100 205137
- [28] Shamblin J, Heres M, Zhou H, Sangoro J, Lang M, Neuefeind J, Alonso J A and Johnston S 2018 Nat. Commun. 986
- [29] Bisogni V, Catalano S, Green R J, Gibert M, Scherwitzl R, Huang Y, Strocov V N, Zubko P, Balandeh S and Triscone J M 2016 Nat. Commun. 7 1
- [30] García-Muñoz J L, Rodríguez-Carvajal J and Lacorre P 1994 Phys. Rev. B 50 978
- [31] García-Muñoz J L, Aranda M A G, Alonso J A and Martínez-Lope M J 2009 Phys. Rev. B 79 1
- [32] Fernández-Díaz M T, Alonso J A, Martínez-Lope M J, Casais M T and García-Muñoz J L 2001 Phys. Rev. B 64 1444171
- [33] Muñoz A, Alonso J A, Martínez-Lope M J and Fernández-Díaz M T 2009 J. Solid State Chem. 182 1982
- [34] Fauth F, Boer R, Gil-Ortiz F, Popescu C, Vallcorba O, Peral I, Fullà D, Benach J and Juanhuix J 2015 Eur. Phys. J. Plus. 130 160
- [35] Brown I D and Altermatt D 1985 Acta Crystallogr. B 41 244
- [36] Causa M T, Sanchez R D, Tovar M, Alonso J A and Martinez-Lope M J 2003 Phys. Rev. B 68 24429
- [37] García-Muñoz J L, Rodríguez-Carvajal J and Lacorre P 1992 Europhysics Lett. 20 241